# Electrical properties of silicon substituted Co–Zn Spinel Ferrite

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**Abstract-** Spinel ferrites with general formula  $Co_{0.7+x} Zn_{0.3} Si_x Fe_{2-2x} O_4$  (for x = 0.0 to 0.4) were prepared by standard double sintering ceramic method. Electrical properties of the present spinel system were studied by means DC and AC resistivity measurements. Both DC resistivity  $\rho_{dc}$  and AC resistivity  $\rho_{ac}$  are found to increase with increase in Si concentration from x = 0.0 to 0.1 and then they decrease with increase in Si concentration from x = 0.2 to 0.4. AC resistivity  $\rho_{ac}$  of all the samples is found to increase at lower temperature and then it decreases with increase in temperature.

Keywords: spinel ferrite, Electrical Properties, hoping

### 1) INTRODUCTION

Ferrites have wide range of applications due to their interesting electrical and dielectric properties. The electrical properties of ferrites depend upon chemical composition, Methods of preparation and sintering temperature [1,2].

The electrical conductivity in ferrites can be explained on the basis of Verwav mechanism [3]. The study of dielectric properties of ferrites produces valuable information on the behavior of electronic carriers charge leading to greater understanding of the mechanism of dielectric polarization. It is reported that incorporation of Si in CoFe<sub>2</sub>O<sub>4</sub> necessitates adjustment of both Co<sup>2+</sup> and Fe<sup>3+</sup> ions [4]. The present work reports effect of Si substitution on DC and AC resistivities of Co-Zn ferrite.

#### 2) EXPERIMENTAL

Five samples of the ferrite system  $Co_{0.7+x}$ Zn<sub>0.3</sub> Si<sub>x</sub> Fe<sub>2-2x</sub> O<sub>4</sub> (for x = 0.0 to 0.4) were prepared by the double sintering ceramic method using AR grade oxides CoO, ZnO, SiO<sub>2</sub> and Fe<sub>2</sub>O<sub>3</sub>. The XRD patterns of all the samples were recorded using CuK $\alpha$ radiation on a Pw710 diffractometer.

The electrical resistivity as a function of Si concentration was measured using a two probe method for all samples in the form of pellets. AC electrical resistivity, as a function of temperature was measured using Aplab made LCR-Q meter at fixed frequency of 1KHz.

### 3) RESULTS AND DISCUSSION 3.1 XRD ANALYSIS

Analysis of the XRD patterns revealed that all the samples have a single phase cubic structure. Comparing the calculated and observed X-ray intensity ratios  $I_{220}/I_{400}$  and  $I_{400}/I_{422}$  and using site preference of various ions, the cation distribution of the present system can be written as

For x = 0.0  $(Zn_{0.3} \text{ Fe}_{0.7})^{\text{A}} [Co_{0.7} \text{ Fe}_{1.3}]^{\text{B}}$ For x=0.1 to 0.4  $(Co_{0.05} Zn_{0.3} \text{ Si}_{0.05+y} \text{ Fe}_{0.6-y})^{\text{A}} [Co_{0.75+y} \text{ Si}_{0.05} \text{ Fe}_{1.2-y}]^{\text{B}}$ 

Where x = 0.1 + y, y = 0.0, 0.1, 0.2, 0.3

## **3.2 DC RESISTIVITY**

The values of DC resistivity ( $\rho_{dc}$ ) of the present system with varying Si concentration at fixed temperature of 500 K are shown in Table-1. It is evident that dc resistivity increase with increases in Si concentration from x = 0.0 to 0.1 and then it decreases with increase in x form x = 0.2 to 0.4. Due to complex structural composition of these ferrites, the conduction mechanism can be attributed to a combination of several processes. Fe<sup>2+</sup> ions are formed due to partial reduction of small fraction of Fe<sup>3+</sup> to Fe<sup>2+</sup> ions due to volatilization of Zn elevated firing temperature during at sintering process [5]. In the present system, 2Fe3+ ions are replaced by Si4+ ions and Co<sup>2+</sup> ions. Also CoO has an affinity for oxidation resulting in the formation of Co<sup>3+</sup> ions. So increasing resistivity of the system

with increase of Si content up to x = 0.1may be due to electron – hole compensation.

The decrease in dc resistivity  $\rho_{dc}$  with increase of Si content from x = 0.2 to 0.4 may be attributed to the presence of larger number of cobalt ions on B-site, which favor the hoping mechanism between Co<sup>2+</sup> and Co<sup>3+</sup> ions. Thus P-type i.e. hole hoping between Co<sup>2+</sup> and Co<sup>3+</sup> ions is predominant conduction mechanism in the present system containing higher concentration of cobalt. These observations are consistent with the earlier reported results [6,7] according to which the presence of Co<sup>2+</sup> ions on B site lowers the resistivity.

## 3.3 AC RESISTIVITY

Table- 1 shows variation of AC resistivity  $\rho_{ac}$  with variation of Si concentration. Similar to DC resistivity, AC resistivity is found to increase with increase in Si concentration from x = 0.0 to 0.1 and then it decreases for higher concentration of Si. Only difference is that the AC resistivity values are smaller than DC resistivity values.

Fig.1 shows plots of log  $\rho_{ac}$  against reciprocal of temperature. It is evident that for all samples of the present system,  $\rho_{ac}$  is found to increase initially with temperature (up to about 350K) and then the characteristic variation of  $\rho_{ac}$  follows the typical semiconducting behavior i.e. it decreases with increase in temperature. The anomalous variation of  $\rho_{ac}$  resistivity at lower temperature (up to 350K) may be due to ionic drift current. The variation of  $\rho_{ac}$  with temperature can be attributed to current due to electrons in conduction band, current due to electron hoping and ionic drift current [8]. results indicate The present that contribution from ionic drift current is more predominant in the anomalous region of  $\rho_{ac}$ 

plots. The ionic contribution becomes negligible at higher temperatures and  $\rho_{ac}$  resistivity becomes entirely electronic.

# CONCLUSION

- 1) Increase in dc resistivity up to Si content x = 0.1 is due to electron hole compensation.
- Decrease in dc resistivity for Si content x ≥ 0.2 is due to hoping mechanism between Co<sup>2+</sup> and Co<sup>3+</sup> ions.
- 3) Anomalous variation of ac resistivity at lower temperature is due to ionic drift current.

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## REFERENCES

- B.P. Ladgaonkar, C.B. Kolekar, P. N.
   Vasambekar, A.S.Vaingankar, Ind.
   J.Eng. Mater. Sci. 7 (2000) 419.
- [2] A.S. Vaingankar, S.G. Kulkarni, M.S. Sagare J. Phys. France 7 (1997) C 1 – 155.
- [3] E.J.W. Verwey, P.W. Haaijmann, E.C. Romeyn, G.M.Van Oosterhont, Philips Res. Rep. 5 (1950) 173.
- [4] S.S. Shinde, K.M. Jadhav, Mater Lett. 37(1998) 63-67.
- [5] Van Uitert L.C. Proc. IRE (USA), 1294 (1956)1303.
- [6] A.B. Devale, D.K. Kulkarni, J. Pure and Appl. Phys. 16(1978) 697.
- [7] R. Satyanarayana, S.R. Murthy, T.S. Rao and M.D. Rao, J. Less. Commun. Metals, 86 (1982) 115.
- [8] R.H.J. Waldron, Appi. Phys. (USA), 43(1972)1186.

Composition	DC resistivity	AC resistivity
X	$ ho_{dc}$	$ ho_{ac}$
0.0	1.31 x 10 <sup>6</sup>	1.26 x 10 <sup>5</sup>
0.1	3.53 x 10 <sup>6</sup>	6.31 x 10 <sup>5</sup>
0.2	1.61 x 10 <sup>6</sup>	5.01 x 10 <sup>5</sup>
0.3	0.34 x 10 <sup>6</sup>	1.26 x 10⁵
0.4	0.07 x 10 <sup>6</sup>	0.63 x 10⁵

Table 1-DC and AC resistivity of Co<sub>0.7+x</sub> Zn<sub>0.3</sub> Si<sub>x</sub> Fe<sub>2-2x</sub> O<sub>4</sub>



Fig. 1- Variation of log  $\rho_{ac}$  with reciprocal of temperature